## organic papers

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#### Key indicators

Single-crystal X-ray study T = 173 K Mean  $\sigma$ (C–C) = 0.002 Å R factor = 0.041 wR factor = 0.115 Data-to-parameter ratio = 17.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 2,2'-[(3-Hydroxyphenyl)methylene]bis(3-hydroxy-5,5dimethyl-2-cyclohexen-1-one) hydrate

The title compound is a bis-dimedone derivative which crystallizes as a monohydrate,  $C_{23}H_{28}O_5 \cdot H_2O$ . Its crystal packing is stabilized by hydrogen bonds involving water.

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## Comment

Some years ago, we accidentally encountered the twinned crystal structures of two similar compounds bearing the bisdimedone moiety: bis(2-hydroxy-4,4-dimethyl-6-oxo-1-cyclohexenyl)phenylmethane (Bolte *et al.*, 1997) and 2,2'-methylenebis(3-hydroxy-5,5-dimethylcyclohex-2-en-1-one) (Bolte & Scholtyssik, 1997). Since these compounds can easily be prepared (Hünig *et al.*, 1979), we have decided to synthesize various derivatives with the objective to find out if these are also twinned (Bolte *et al.*, 2001). We hoped to find more twinned crystal structures in order to understand why some of these simple compounds containing the common bis-dimedone skeleton are twinned. Unfortunately, none of these, including the title compound, (I), turned out to be twinned.



The two cyclohexenone rings display envelope conformations with C14 and C24 deviating by 0.644 (2) and 0.653 (2) Å, respectively, from the plane of the remaining five ring atoms. Both apices are extended towards the phenyl ring. Two strong intramolecular hydrogen bonds are formed connecting the two cyclohexenone rings. The water molecule bridges two different molecules *via* its H atoms and acts as an acceptor for the H atom of the hydroxyl group of a third molecule.

#### Experimental

The title compound was synthesized according to Nagarajan & Shenoy (1992).

Crystal data  $D_x = 1.221 \text{ Mg m}^{-3}$ C23H28O5·H2O  $M_r = 402.47$ Mo  $K\alpha$  radiation Monoclinic,  $P2_1/n$ Cell parameters from 8192 a = 8.948(1) Å reflections b = 27.128 (3) Å  $\theta = 0-25^{\circ}$  $\mu=0.09~\mathrm{mm}^{-1}$ c = 9.045(1) Å  $\beta = 94.30 (1)^{\circ}$ T = 173 (2) K V = 2189.4 (4) Å<sup>3</sup> Block colourless Z = 4 $0.70 \times 0.65 \times 0.45 \text{ mm}$ 

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#### Data collection

Siemens CCD three-circle diffractometer  $\omega$  scans Absorption correction: empirical (SADABS; Sheldrick, 1996)  $T_{\min} = 0.941, T_{\max} = 0.962$ 41 555 measured reflections 4836 independent reflections 4088 reflections with  $I > 2\sigma(I)$ 

#### Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.041$   $wR(F^2) = 0.115$  S = 1.014836 reflections 283 parameters H atoms treated by a mixture of independent and constrained refinement

Table 1

Hydrogen-bonding geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D - H \cdot \cdot \cdot A$
O12−H12···O26	1.09 (3)	1.49 (3)	2.560(1)	164 (2)
O22-H22···O16	1.05 (3)	1.62 (3)	2.664 (1)	171 (3)
$O1W-H1W \cdots O16$	0.92 (2)	1.88 (2)	2.795 (2)	172 (2)
$O33-H33\cdots O1W^i$	0.92 (2)	1.76 (2)	2.684 (2)	174 (2)
$O1W - H2W \cdot \cdot \cdot O12^{ii}$	0.97 (3)	1.90 (3)	2.870 (2)	175 (2)

 $R_{\rm int} = 0.028$ 

 $\theta_{\max} = 27.1^{\circ}$  $h = -11 \rightarrow 11$ 

 $k = -34 \rightarrow 34$ 

 $l = -11 \rightarrow 11$ 

486 standard reflections

frequency: 1200 min

intensity decay: none

 $w = 1/[\sigma^2(F_o^2) + (0.0615P)^2]$ 

Extinction correction: SHELXL97

Extinction coefficient: 0.0036 (10)

+ 0.8381*P*] where  $P = (F_o^2 + 2F_c^2)/3$ 

 $\begin{array}{l} (\Delta/\sigma)_{\rm max} = 0.001 \\ \Delta\rho_{\rm max} = 0.31 \ {\rm e} \ {\rm \AA}^{-3} \\ \Delta\rho_{\rm min} = -0.17 \ {\rm e} \ {\rm \AA}^{-3} \end{array}$ 

Symmetry codes: (i) 1 - x, -y, 1 - z; (ii) 1 + x, y, z.

All H atoms bonded to C atoms were located by difference Fourier synthesis and refined with fixed individual displacement parameters  $[U(H) = 1.5U_{eq}(C_{methyl}) \text{ or } U(H) = 1.2U_{eq}(C)]$  using a riding model with C-H(aromatic) = 0.95, C-H(methyl) = 0.98, C-H(secondary) = 0.99 or C-H(tertiary) = 1.00 Å. All hydroxyl H atoms were refined isotropically.

Data collection: *SMART* (Siemens, 1995); cell refinement: *SMART*; data reduction: *SAINT* (Siemens, 1995); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1990); program(s) used to



#### Figure 1

Perspective view of the title compound with the atom numbering; displacement ellipsoids are at the 50% probability level.

refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *XP* in *SHELXTL-Plus* (Sheldrick, 1991).

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